C 40514	(Pages: 2)	Name
		Rog No

## SIXTH SEMESTER U.G. DEGREE EXAMINATION MARCH 2023

(CBCSS—UG)

Chemistry/Polymer Chemistry
CHE 6B 09—INORGANIC CHEMISTRY—IV

(2019 Admission onwards)

Time: Two Hours

Maximum: 60 Marks

## Section A (Short Answers)

Answer all questions.

Each question carries 2 mark.

Ceiling 20.

- 1. What is the difference between DTA and DSC?
- 2. What is the use of thermogravimetric analysis?
- 3. Cupric salts are coloured while cuprous salts are colourless. Give reason.
- 4. Write the formula for spin only magnetic moment of transition ions based on number of unpaired electrons.
- 5. Why is the increase in the first ionization energy of transition elements not vary regularly with an increase in atomic number?
- 6. What is the difference between labile and inert complexes?
- 7. What is spectrochemical series?
- 8. While  $Co[(H_2O)_6]^{2+}$  is pink in colour,  $Co(Cl)_4]^{2-}$  is blue in colour. Why?
- 9. Draw the shapes of Fe  $(CO)_5$  and Ni  $(CO)_4$ .
- 10. What is Zeise's salt? Write its structure.
- 11. Write a note on effect of mercury on living body.
- 12. How does Hemoglobin differ from myoglobin?

(Ceiling of marks: 20)

Turn over

2 C 40514

## Section B (Paragraph)

Answer all questions.

Each question carries 5 marks.

Ceiling 30.

- 13. Differentiate between SEM and TEM.
- 14. Why do lanthanoids form coloured complexes?
- 15. What is lanthanide contraction? What are its consequences?
- 16. What are actinides? Why are they so called?
- 17. Cobalt (III) easily forms low spin complexes whereas Cobalt (II) does not. Explain.
- 18. Give an account of the classification of organometallic compounds by nature of bonding.
- 19. Give an account of the biological significance of Cobalt in living systems

(Ceiling of marks: 30)

## Section C (Essay)

Answer any one questions.

The question carries 10 marks.

- 20. (a) Describe briefly the general characteristics of the f block elements in the periodic table with emphasis on their electronic configuration,
  - (b) Comment on the industrial importance of Lanthanides.

Any 10 points full mark.

- 21. (a) Write in detail the preparation and properties of Ferrocene.
  - (b) Discuss the nature of bonding in metal carbonyls.

 $(1 \times 10 = 10 \text{ marks})$